

Versiliaite and apuanite: derivative structures related to schafarzikite

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Abstract

Schafarzikite, FeSb_2O_4 , is tetragonal, space group $P4_2/mbc$, with $a = 8.59$, $c = 5.91\text{\AA}$. The crystal structure is characterized by the presence of edge-sharing iron octahedra connected with corner-sharing antimony ψ -tetrahedra (Fischer and Pertlik, 1975).

Versiliaite is orthorhombic, space group $Pbam$, $a = 8.492$, $b = 8.326$, $c = 11.938\text{\AA}$. Its crystal structure is related to that of schafarzikite in the following manner: every fourth Sb^{3+} ion in the ψ -tetrahedral chains is substituted by a Fe^{3+} ion and a sulphide anion is added between two Fe^{3+} cations in adjacent chains. The corner-sharing Fe^{3+} tetrahedra connect the chains to build double-chain ribbons. The charge balance, altered by the insertion of sulphide anions, is restored by the substitution of Fe^{2+} cations in the octahedral chains by Fe^{3+} cations.

Apuanite is tetragonal, space group $P4_2/mbc$, $a = 8.372$, $c = 17.974\text{\AA}$. Its crystal structure can be derived from that of schafarzikite by substituting every third Sb^{3+} ion in the ψ -tetrahedral chains with an Fe^{3+} cation and adding sulphide anions which connect adjacent chains to build infinite ladders. As in versiliaite the charge balance is restored by substitution of Fe^{2+} cations in octahedral chains with Fe^{3+} cations.

The crystal chemistry of these minerals is clarified and the vacancies in sulphide sites as well as the incompleteness in cationic substitutions are explained. The lines for a systematic derivation of possible, yet unknown, related structures are developed.

Introduction

Versiliaite and apuanite were described by Mellini *et al.* (1979); they showed the close chemical, physical, and crystallographic relationships of the two minerals with schafarzikite, FeSb_2O_4 , from which versiliaite and apuanite can be derived by a concerted mechanism of insertion, substitution, and oxidation.

The aim of this paper is to describe and discuss the main features of their crystal structures and to develop a comprehensive crystal chemistry for the schafarzikite group of minerals.

Structure determination.

The experimental parameters characterizing the intensity data collection step are summarized in Table 1. Two equivalent sets were collected for both crystals; the diffraction intensities were corrected for the usual geometrical factors, as well as for absorption by the semiempirical method of North *et al.* (1968); symmetry-related reflections were thereafter checked for equivalence and averaged to pro-

duce the unique sets. In the subsequent structure-factor calculations, all the scattering factors for neutral atoms and anomalous dispersion corrections were taken from the International Tables for X-ray Crystallography, or calculated from them when a hybrid curve was used.

Versiliaite

The structure determination was undertaken in the centrosymmetric space group $Pbam$, indicated by the statistical averages and distribution of E_{hkl} values; the choice appeared correct in the subsequent refinement stage. A trial model was built, starting from the parameters given by Fischer and Pertlik (1975) for schafarzikite and taking account of the doubled c parameter. The use of the scattering factors of antimony for atoms at $z = 0.0$ and $z = 0.25$ and of arsenic for an atom at $z = 0.5$ led to the doubling of the c translation. Moreover we used the full reflections set, including the weakest ones corresponding to $l = 2n + 1$. The refinement was led by Fourier syntheses and least-squares cycles for all the three-coordinated and

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7	922	-929	10	59	-50	9	45	-25	1	63	-85	H,4,2	8	110	91	H,8,2	2	1295	-1147	5	1436	-1431
8	3283	3284	H,6,0	11	191	191	H,5,1	2	20	-12	H,4,2	8	110	91	H,8,2	2	1295	-1147	5	1436	-1431	
9	453	461	H,2,1	6	28	50	4	4	0	-14	4	53	-98	H,0,3	3	3760	3770	10	3760	3770		
10	1335	1336	H,2,1	6	28	50	4	4	0	-14	4	53	-98	H,0,3	3	3760	3770	10	3760	3770		
11	0	23	H,2,1	6	28	50	4	4	0	-14	4	53	-98	H,0,3	3	3760	3770	10	3760	3770		

6	215	-218	9	216	202	0	1793	-1656	3	1401	-1319	H,7,4	7	688	-695	7	303	246	6	215	-218
8	131	13510	4	11	304	2	1270	1106	4	51	-38	H,7,4	7	688	-695	7	303	246	6	215	-218
10	142	-9611	4	1772	1734	5	840	763	8	0	-114	9	88	-24	9	372	-339	11	104	-89	
11	0	23	6	1583	-1589	6	26	-36	9	727	74310	11	613	640	H,7,5	5	5220	6011	5	3694	3730

2	5220	6011	5	3694	3730	10	1194	1179	8	78	111	H,8,4	4	286	300	8	815	-832	2	5220	6011	
3	165	-140	6	388	411	9	1225	-1228	10	0	47	8	157	77	4	286	300	8	815	-832		
4	5506	-5921	7	3605	-3647	11	579	639	H,0,5	5	217	223	H,0,6	6	104	-89	8	104	-89	3	165	-140
5	104	-89	8	198	-174	1	124	-118	H,0,5	5	217	223	H,0,6	6	104	-89	8	104	-89	3	165	-140

2	5220	6011	5	3694	3730	10	1194	1179	8	78	111	H,8,4	4	286	300	8	815	-832	2	5220	6011	
3	165	-140	6	388	411	9	1225	-1228	10	0	47	8	157	77	4	286	300	8	815	-832		
4	5506	-5921	7	3605	-3647	11	579	639	H,0,5	5	217	223	H,0,6	6	104	-89	8	104	-89	3	165	-140
5	104	-89	8	198	-174	1	124	-118	H,0,5	5	217	223	H,0,6	6	104	-89	8	104	-89	3	165	-140

2	5220	6011	5	3694	3730	10	1194	1179	8	78	111	H,8,4	4	286	300	8	815	-832	2	5220	6011	
3	165	-140	6	388	411	9	1225	-1228	10	0	47	8	157	77	4	286	300	8	815	-832		
4	5506	-5921	7	3605	-3647	11	579	639	H,0,5	5	217	223	H,0,6	6	104	-89	8	104	-89	3	165	-140
5	104	-89	8	198	-174	1	124	-118	H,0,5	5	217	223	H,0,6	6	104	-89	8	104	-89	3	165	-140

7 3226-3257 4 1374 1313 6 1139 1130
8 191-195 5 217 -176 8 279 -281 3
9 810 785 6 91 -24 9 92 45 4
10 22 -56 7 368 -340 10 240 -224 5
H_{5,9} 8 151 136
9 893 896 H_{5,10} 7
10 124 -155 5 177 50 9
6 509 -482 H_{2,10} 6 57 -910
7 62 52 7 627 -615
8 1999-2035 2 0 46 8 61 4
9 216 -159 3 81 -30 9 858 930
H_{6,9} 4 396 -401
5 86 118 H_{6,10} 3
7 633 638 6 941 918
8 317 346 7 90 15 6 1268-1267 6
9 314 -278 8 0 -61 7 0 51 7
H_{7,9} 9 116 -124 8 679 770 8
10 379 -320 9 0 -60 9 356 -290 4
8 1636 1728 H_{3,10} H_{7,10} 10 31 65 5556 5764 9 1900 1961

H_{0,10} 3 1815-1763 7 87 -142
0 1775-1659 5 893 911
2 1090 974 6 0 12
4 1515 1499 7 1307 1303 H_{0,11} 5
6 1591-1608 8 110 79 2 84 74 8 104 8 104 3 2272-3352 8 1685 1661 7 117 -153
8 771 705 9 1427-1457 4 197 -229 9 0 44 2 521 -531 9 200 -186 8 1909-1938
10 1131 109310 22 -4 6 169 15810 85 24 5 556 349 H_{5,12} 10 933 1019
H_{1,10} H_{4,10} 9 69 96 6 121 64
1 311 -314 4 347 -353 10 227 -220 5 599 -635 9 2247-2182 6 0 24 H_{3,16} 2 82 -30
H_{2,13} H_{6,14} 7 566 548 5 110 -103 H_{6,15} 4 70 106 H_{2,17} 3 150 -102

4 308 328 3 234 228 H_{5,14} 6 546 553 3 303 -261 4 1415 1445
5 0 5 4 112 124 H_{5,14} 8 97 -80 4 145 157 6 1685-1743
6 107 -110 5 274 289 H_{5,14} 7 2485 2482
7 155 142 6 0 -59 5 341 333 9 711 -674 H_{1,16} 4 379 -375 4 0 -23
8 134 -70 7 126 -117 6 103 92 H_{3,15} 1 326 -293 7 89 -147 7 73 -141
9 0 25 8 119 77 7 238 -227 H_{3,15} 2 93 -26 8 380 -381 8 106 68
H_{4,13} 9 207 225 8 107 -36 H_{3,16} 5 255 -231 H_{5,16} H_{3,17} 3 150 -102

H_{2,14} H_{6,14} 4 333 301 4 63 35
5 1519-1533 H_{2,14} H_{6,14} 5 127 142 6 41 32 5 0 -123 4 119 150
6 75 -120 2 375 346 6 276 -292 7 344 -314 6 57 13 5 116 113
7 967 1007 3 96 104 7 141 146 8 70 39 9 795 813
8 131 -46 4 64 -55 H_{0,15} H_{4,15} H_{2,16} 6 1327-1412 H_{4,16} 8 467 -475 5 164 -145
9 316 260 5 119 -127 6 219 361 H_{4,15} H_{2,16} 7 95 8 7 150 -161
6 219 361 H_{0,15} H_{4,15} H_{2,16} 6 1327-1412 H_{4,16} 8 467 -475 5 164 -145

3 177 50 9 294 349 9 55 -15 9 264 2596
4 57 50 7 51 -77 6 0 15 5 75 -24 7 384 380
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3 357 -353 7 135 106
4 117 112 8 0 58
5 551 563
6 158 -122
7 227 -275
8 49 -2 2 1657-1632 7 1898-1848 8 80 -12
9 356 -290 4 955 -888 8 130 -11610 198 -166
10 31 65 5556 5764 9 1900 1961
H_{3,11} 10 500 411 H_{4,12} 2 1503-1458
H_{1,12} 4 2629 2606 3 197 97
5 197 -156 4 1369 1361
6 199 -166 5 41 17
7 234 236 6 256 249
8 1685 1661 7 117 -153
9 200 -186 8 1909-1938
10 933 1019

5 599 -635 9 2247-2182 6 0 24
6 110 -103
7 767 767
8 511 493 5 361 353
9 2247-2182 6 0 24
10 500 411
H_{1,12} 4 2629 2606 3 197 97
5 197 -156 4 1369 1361
6 199 -166 5 41 17
7 234 236 6 256 249
8 1685 1661 7 117 -153
9 200 -186 8 1909-1938
10 933 1019

5 599 -635 9 2247-2182 6 0 24
6 110 -103
7 767 767
8 511 493 5 361 353
9 2247-2182 6 0 24
10 500 411
H_{1,12} 4 2629 2606 3 197 97
5 197 -156 4 1369 1361
6 199 -166 5 41 17
7 234 236 6 256 249
8 1685 1661 7 117 -153
9 200 -186 8 1909-1938
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5 599 -635 9 2247-2182 6 0 24
6 110 -103
7 767 767
8 511 493 5 361 353
9 2247-2182 6 0 24
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H_{1,12} 4 2629 2606 3 197 97
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6 199 -166 5 41 17
7 234 236 6 256 249
8 1685 1661 7 117 -153
9 200 -186 8 1909-1938
10 933 1019

8	478	375																			
7	52	-119																			
6	2331	-2317																			
5	363	-381																			
4	454	-329																			
3	379	379																			
2	171	175																			
			H _{2,18}																		
8	89	137																			
7	1328	1323																			
6	57	-84																			
5	2471	2448																			
4	259	246																			
3	565	-576																			
2	505	-486																			
1	2129	2105																			
			H _{1,18}																		
8	2329	-2259																			
7	1962	1989																			
6	2583	-2530																			
5	3068	-3063																			
4	3192	3220																			
3	4927	5063																			
2	488	43																			
1	2105	135																			
			H _{1,18}																		
8	89	137																			
7	1328	1323																			
6	57	-84																			
5	2471	2448																			
4	259	246																			
3	565	-576																			
2	505	-486																			
1	2129	2105																			
			H _{1,18}																		
8	2329	-2259																			
7	1962	1989																			
6	2583	-2530																			
5	3068	-3063																			
4	3192	3220																			
3	4927	5063																			
2	488	43																			
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			H _{1,18}																		
8	89	137																			
7	1328	1323																			
6	57	-84																			
5	2471	2448																			
4	259	246																			
3	565	-576																			
2	505	-486																			
1	2129	2105																			
			H _{1,18}																		

4 1203 1270

3 90 56 6 174 138

2 1348-1364 4 104 87 5

5 64 69

4 292 296

5 1864 1886

2 1370 1853

3 333 329